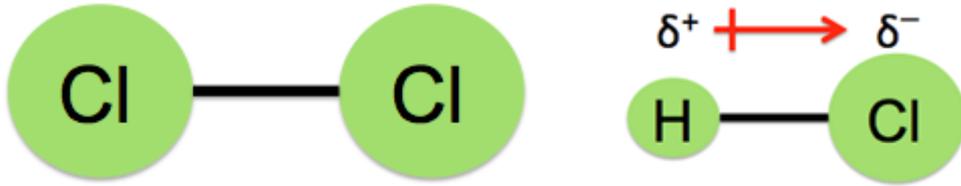
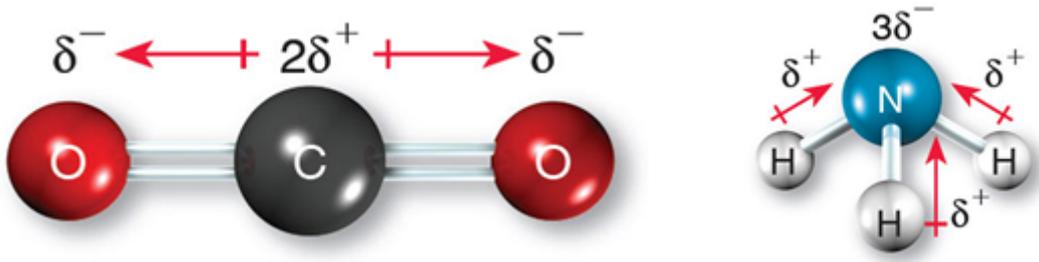


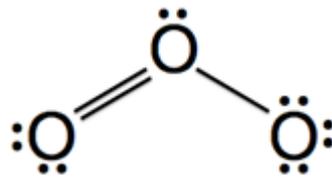
1. 2 atoms of CO_2 and 2 atoms of NH_3 are shown. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal. The CO_2 molecule has a central carbon atom bonded to two oxygen atoms. The NH_3 molecule has a central nitrogen atom bonded to three hydrogen atoms. The CO_2 molecule is shown with partial charges δ^- on the oxygen atoms and $2\delta^+$ on the carbon atom. The NH_3 molecule is shown with a partial charge $3\delta^-$ on the nitrogen atom and δ^+ on each hydrogen atom.



1. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.



2. CO_2 and NH_3 are shown. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.



3. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.

The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.

The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.

(1) The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.



4. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.

A. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal. B. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal. C. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal. D. The CO_2 molecule is linear and the NH_3 molecule is trigonal pyramidal.

