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hueiying.ho@gmail.com

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1959
Richard Feynman
There's Plenty of Room at the Bottom

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2014a K-2 K-12 (ISBN 978-986-04-1575-9)

2014b K-2 K-12 (ISBN 978-986-04-1576-6)

2014c K-12 (ISBN 978-986-04-1582-7)

2014d K-12 (ISBN 978-986-04-1581-0)

2016a K-12 (ISBN 978-986-04-5288-4)

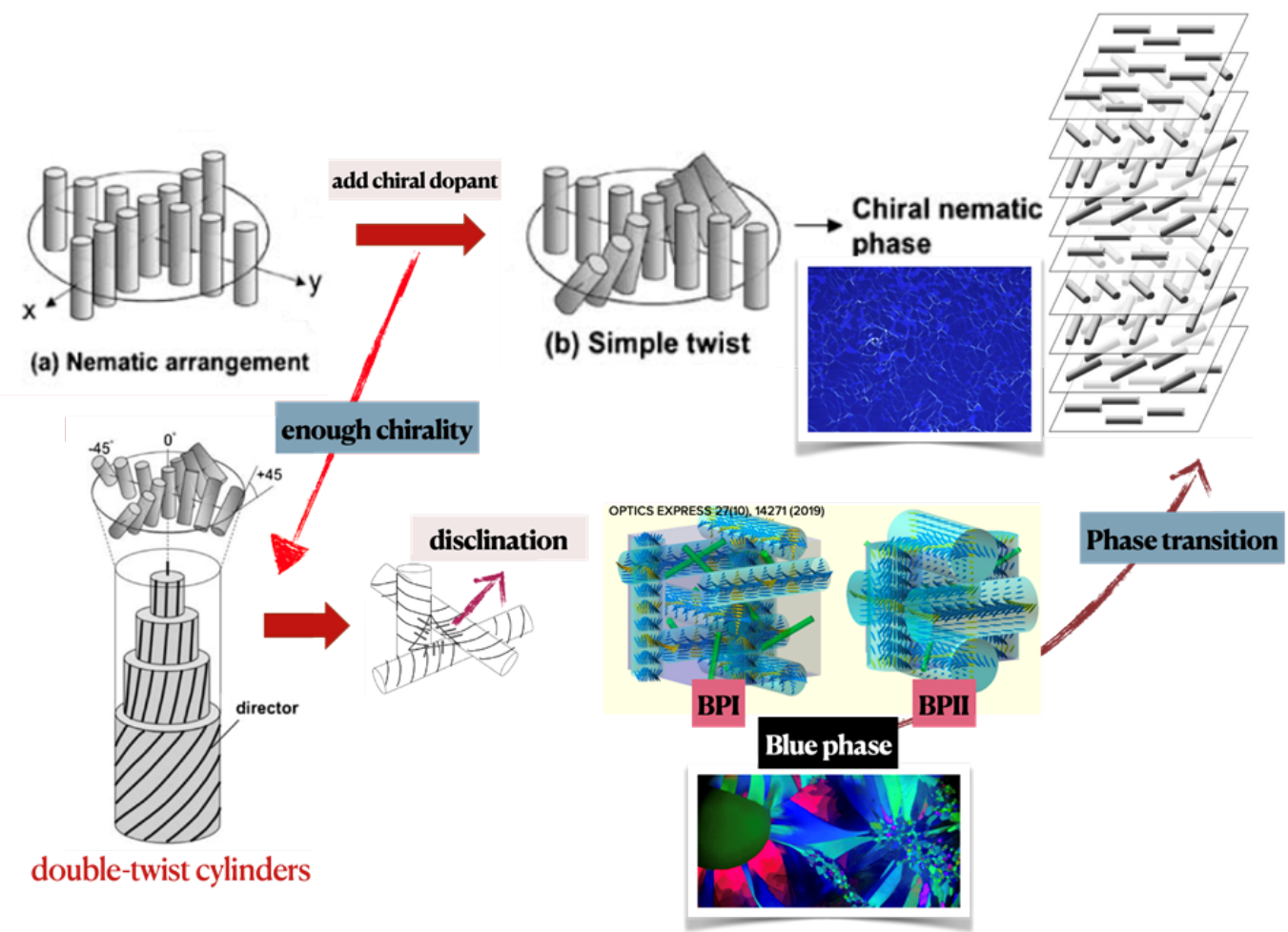
2016b K-12 (ISBN 978-986-04-5287-7)

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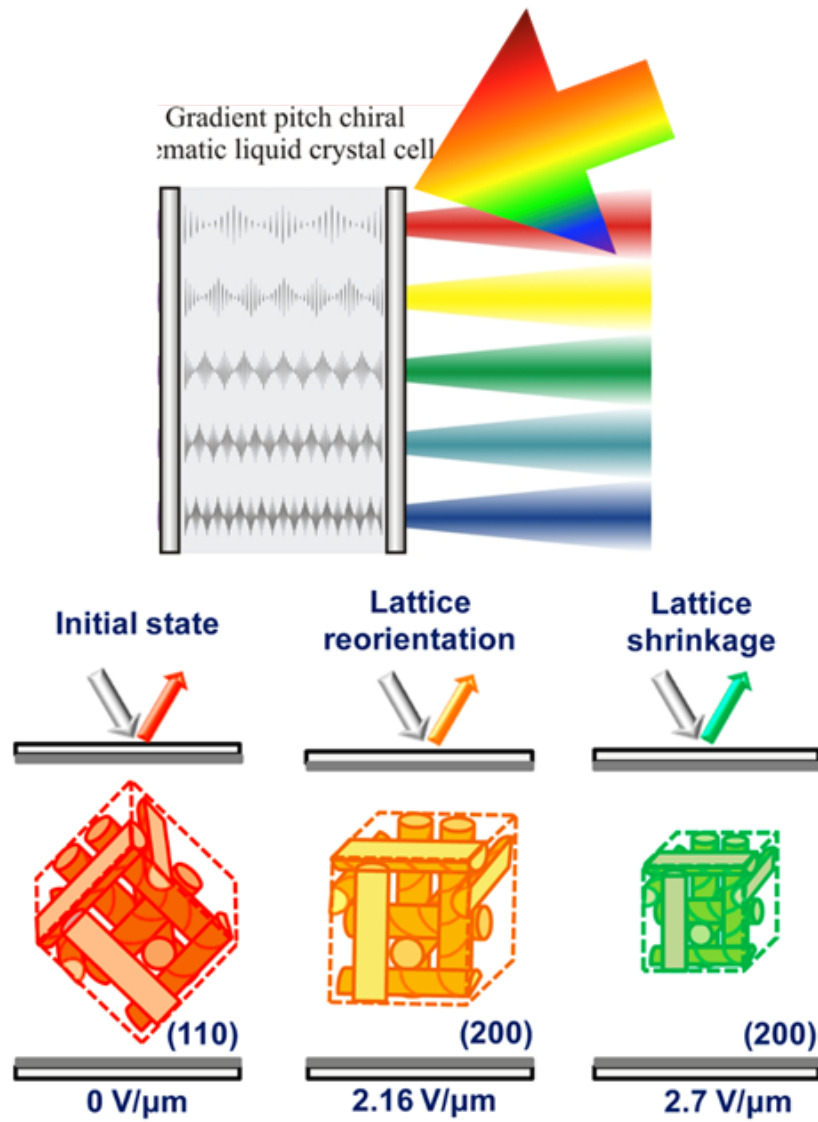
2016d K-12

liquid crystal) (blue phase) 2007

(double-twist cylinder, DTC) DTC $\pm 45^\circ$



(a) Nematic LC (b) chiral matter chiral nematic LC (double-twist cylinder)



(a)

(b)

5 (a) Photo credit: University of Cambridge
 (b) Photo from: ACS Appl. Mater. Interfaces 2017, 9, 39569-39575

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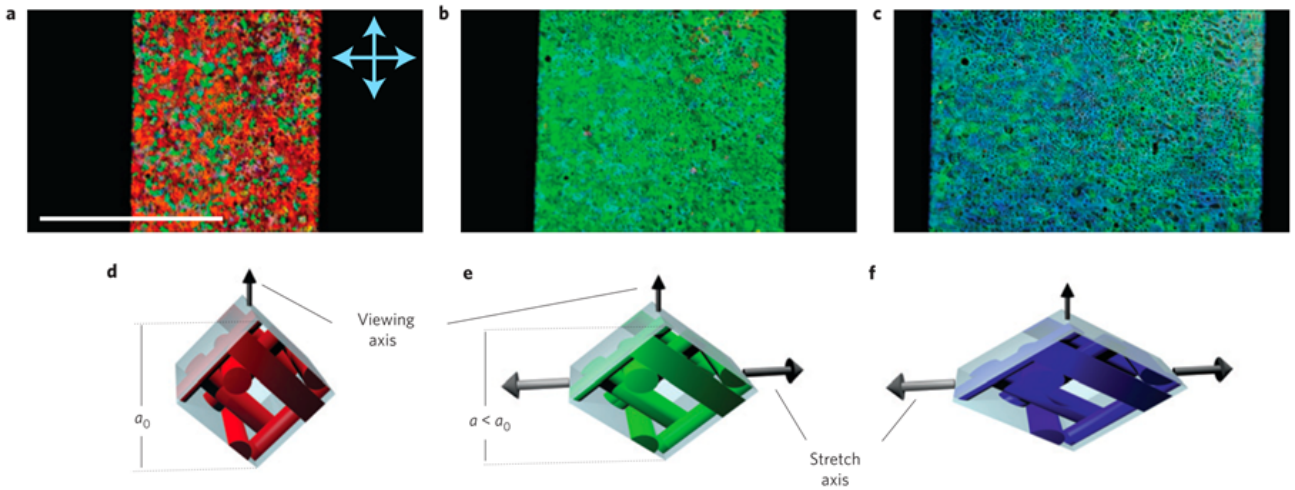


Figure 6. Photo from: NATURE MATERIALS, 13, 817 (2014)

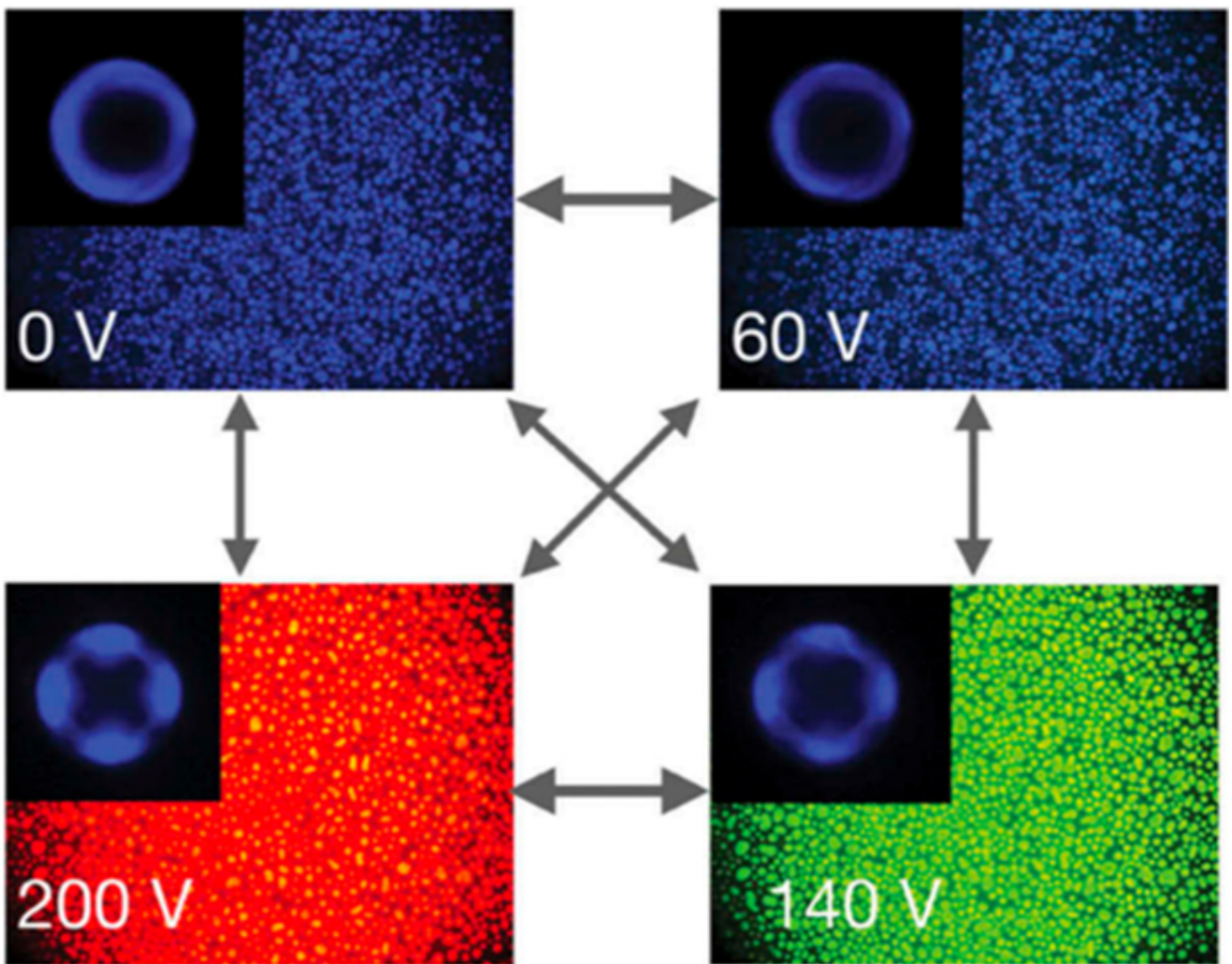


Figure 7. (1, 0, 0) photo from: NATURE MATERIALS, 13, 817 (2014)

Graphene was first synthesized in 1991 by Iijima [1] and was subsequently synthesized by Bianco & Prato, 2003; Davis et al., 2003 [2] and was subsequently synthesized by Bianco & Prato, 2003; Davis et al., 2003 [2].

The synthesis of SWCNT [3], DWCNT [4] and MWCNT [5] was first reported by Iijima in 1991 [1].

■ 石墨烯的氧化

□ 氧化石墨烯

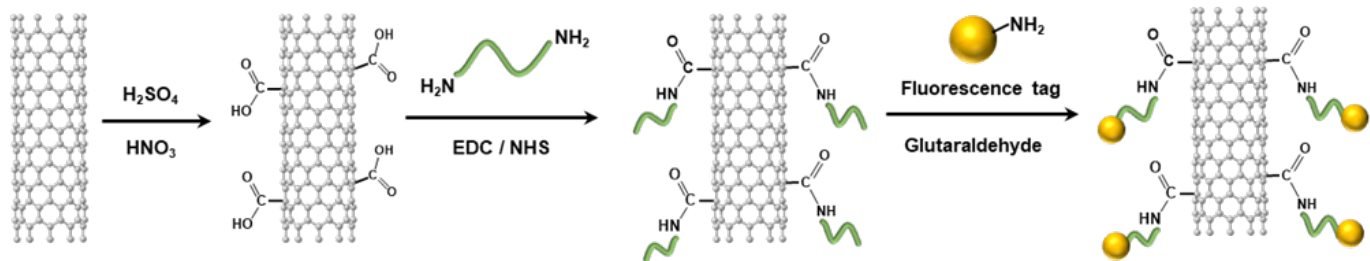
氧化石墨烯 (GO) 是一种由单层石墨烯经氧化反应得到的二维碳材料，其表面含有大量的含氧官能团，如羧基、羟基、环氧基等。这些官能团使得 GO 具有良好的分散性和可加工性，广泛应用于复合材料、传感器、储能材料等领域。

□ a 氧化石墨烯的氧化反应

□ b 氧化石墨烯的还原反应

□ 氧化石墨烯的还原

氧化石墨烯 (GO) 的还原反应是指将 GO 表面的含氧官能团去除，使其恢复为石墨烯的过程。常用的还原剂包括抗坏血酸、水合肼、硼氢化钠等。还原后的材料称为还原氧化石墨烯 (rGO)。

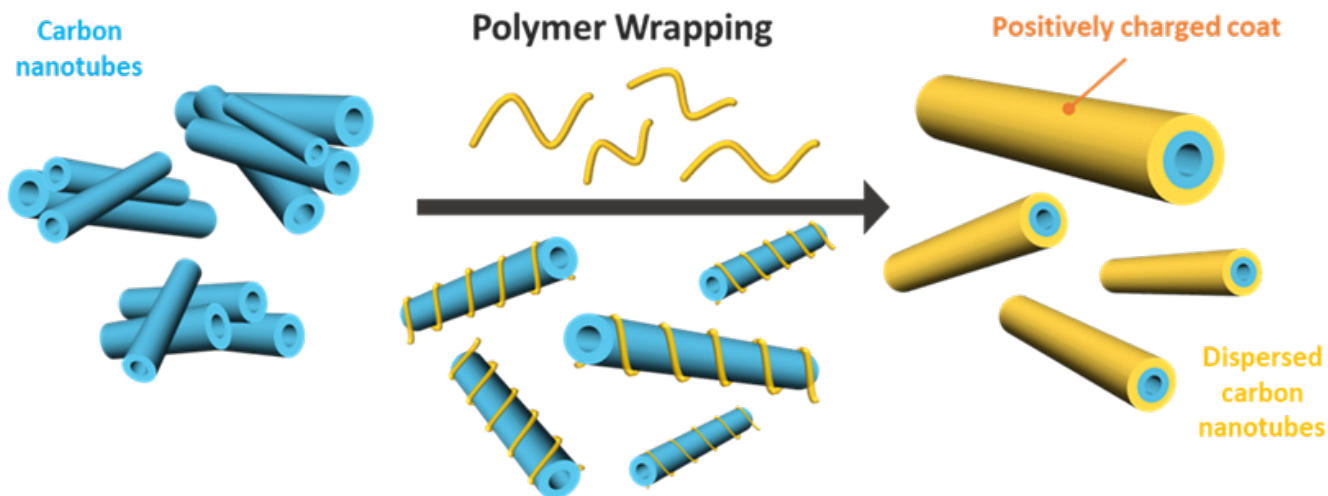


□1 氧化石墨烯的还原

□ 石墨烯的还原

石墨烯的还原是指将氧化石墨烯 (GO) 还原为石墨烯 (G) 的过程。常用的还原剂包括抗坏血酸、水合肼、硼氢化钠等。还原后的材料称为还原氧化石墨烯 (rGO)。

π-π interactions (Mickelson et al., 1998; Pekker et al., 2001; Chen et al., 1998; Kamaras et al., 2003)



2. 分散性向上 (分散性)

分散性向上 (分散性) は、CNT の分散性を向上させるための重要な要素である。分散性向上には、CNT の表面に親水性基を付与する必要がある。これは、CNT の表面に官能基を導入することで実現される。官能基の種類としては、カルボキシル基、ヒドロキシ基、アミノ基などが挙げられる。これらの官能基は、CNT の表面に付着し、水分子と相互作用することで分散性を向上させる。また、分散剤の存在下で CNT を分散させることも有効である。分散剤は、CNT の表面に付着し、CNT と水分子との相互作用を促進することで分散性を向上させる。分散剤の種類としては、表面活性剤、高分子分散剤などが挙げられる。分散剤の濃度や分散時間など、分散条件の最適化も重要な要素である。分散性向上は、CNT の応用範囲を拡大させるための重要な課題である。

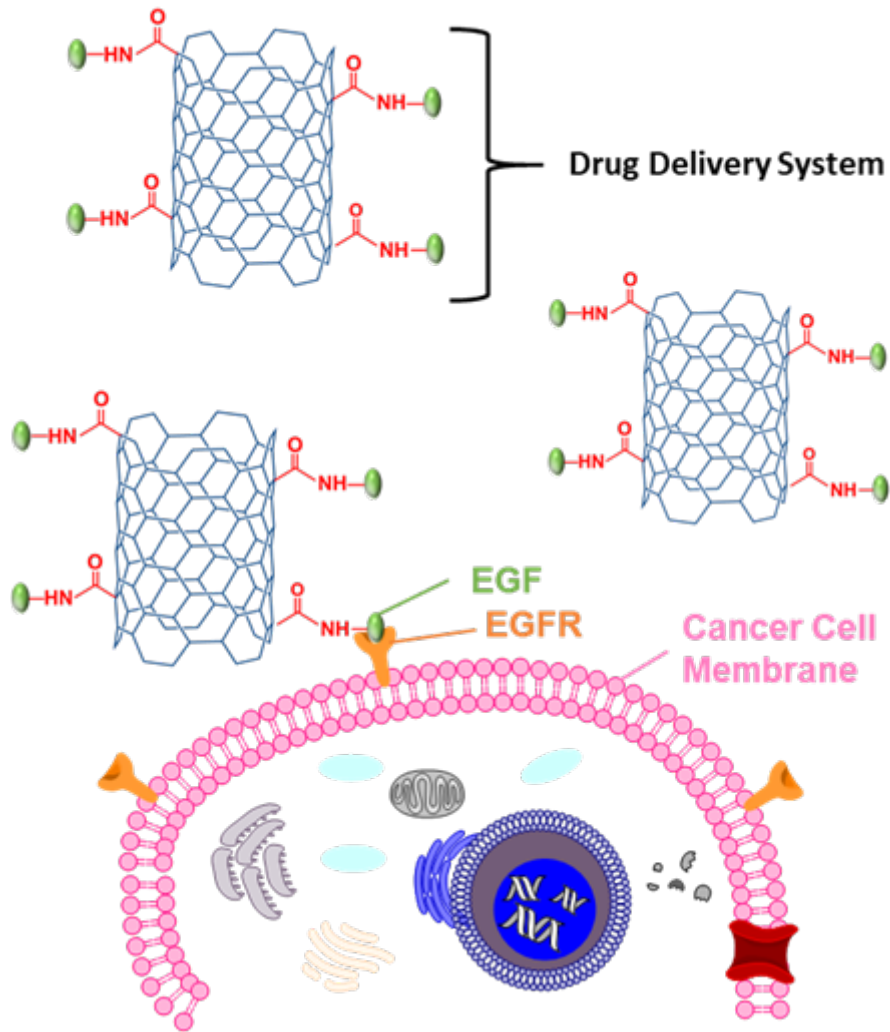
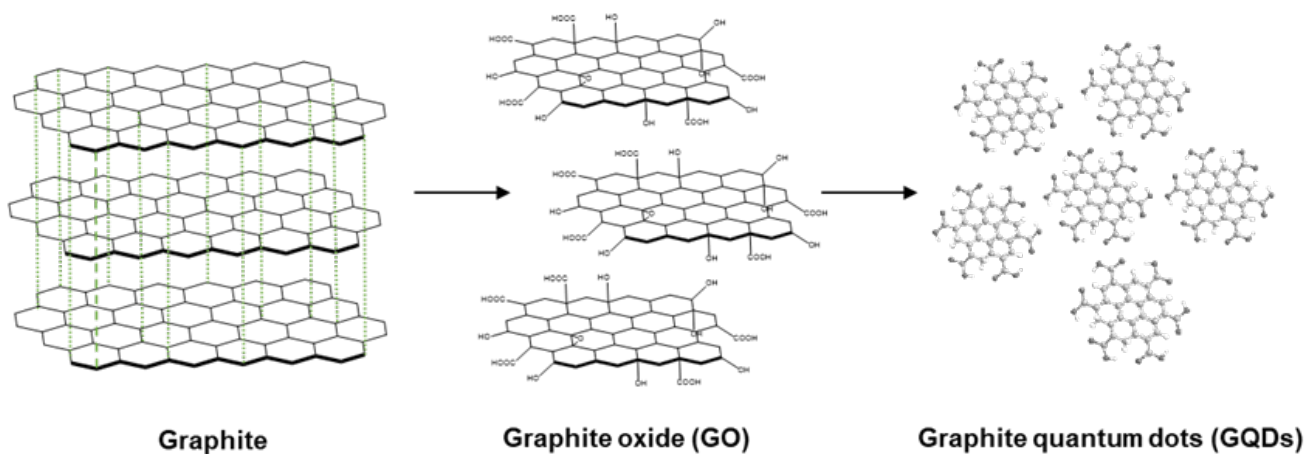


Figure 3. Drug delivery system using a carbon nanotube (CNT) functionalized with amino groups.

Graphene-based nanomaterials

Graphene-based nanomaterials (GNMs) are a class of carbon-based nanomaterials that have attracted significant attention due to their unique properties and potential applications in various fields (Luo et al., 2013; Peng et al., 2012; Shen et al., 2012). GNMs include graphene, graphite oxide (GO), and graphite quantum dots (GQDs). Graphene is a single layer of carbon atoms arranged in a hexagonal lattice. GO is a derivative of graphene with oxygen-containing functional groups attached to the carbon lattice. GQDs are small, single-layered graphene sheets with a diameter of less than 10 nm. GNMs have a large surface area, high electrical conductivity, and excellent mechanical strength. They are used in various applications, including drug delivery, biosensing, and energy storage.



pi-pi interactions (π-π interactions) between the aromatic rings of the DNA base pairs and the aromatic rings of the carbon nanotubes.

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Lo et al., 2020 DNA

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Collagen

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Omega-3

K-12

■ Omega-3

Omega-3, EPA, docosahexaenoic acid, DHA, cytokine, Goldberg & Katz, 2007; Kiecolt-Glaser et al., 2011, Omega-3, Yvonne, 2007, signal protein, Lackie, 2010

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porous biological scaffold, Collagen, 2019, polypeptide, 2014

cross-linking 2019 2019

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K-12 Omega-3 K-12

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2019 12 http://dgnet.com.tw/articleview.php?article_id=18635&issue_id=3629

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chen7329@gmail.com¹

luchowch@tea.ntue.edu.tw²

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Lotus effect 2004 10nm ~100nm 2020

Wandersee et al., 1994 Palmer & Flanagan, 1997 Prediction Observation Explanation POE 2000 Liew, 1995 POE

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主題	教學概念	教學活動內容
疏水性	蓮葉表面與水的接觸角大於90度，具有疏水性	將水滴在蓮葉表面上，觀察~ 1. 蓮葉上的水珠形狀如何？ 2. 蓮葉上的水是滑動的還是滾動的？
自潔作用	滾動的水珠會把灰塵帶走，達到自我潔淨的效果	在蓮葉上灑一些爽身粉，再滴水，觀察~ 1. 水流過有爽身粉的蓮葉時，會產生什麼反應？ 2. 爽身粉會在水珠的裡面還是外面？

階段	教師與學生的角色	學生應答情形
預測 (P)	教師詢問、引導學生設計實驗： 1. 將水滴在未燻黑與燻黑的紙杯底部，水滴會產生什麼情形？ 2. 在未燻黑與燻黑的紙杯底部灑上爽身粉，並各滴一滴水，水和爽身粉又會產生什麼情形？	1. 水滴在燻黑的紙杯底部，會呈現圓形水滴，呈現滾動現象。 2. 水珠會把爽身粉帶走，爽身粉會溶解在水中，並且在水珠裡面。
	學生分別進行實驗觀察與紀錄： 1. 學生觀察並紀錄實驗結果。 2. 學生實驗後發現與預測不相，多次重複實驗、確認實驗結果。	1. 水滴在燻黑的紙杯底部會形成水珠，水珠經過的地方，會把碳顆粒帶走，水珠也會被碳顆粒包圍而變成黑色。 2. 碳顆粒是黏附在水珠外面，不是溶解在水珠裡面。
解釋 (E)	學生比較實驗預測與觀察間的想法，解釋現象及成因： 1. 學生自我比較實驗前後的想、修正原有的想法。 2. 教師引導學生比較、說明燻黑的紙杯杯底表面和蓮葉表面間的關聯性。	1. 紙杯底部燻黑後，杯底會轉變成超疏水性，當水珠滾動便可讓碳黑黏在水珠表面，並將它帶走。 2. 碳黑表面和蓮葉表面一樣具有蓮葉效應，可以形成水珠而保持乾燥，而且水珠會帶走髒汙而保持乾淨。

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國立中央大學 2004 年 K-12 科學教育研討會論文輯
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2000 年科學教育研討會論文輯 8 頁 1-34

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編

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hueiying.ho@gmail.com

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size effect

size effect: the change in material properties due to the reduction in size of the material. This is a key concept in nanotechnology, where the properties of materials can change significantly as their size approaches the nanoscale.

force & interactions

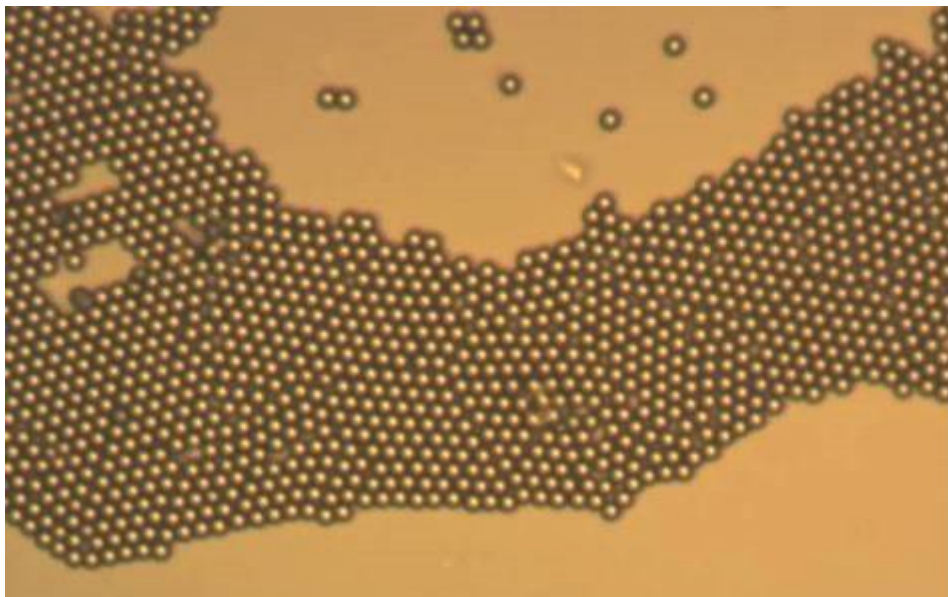
force & interactions: the study of the forces and interactions between particles at the nanoscale. This is a key concept in nanotechnology, where the forces and interactions between particles can be used to design and control nanoscale structures.

quantum effect

quantum effect: the change in material properties due to the quantum mechanical effects of the nanoscale. This is a key concept in nanotechnology, where the quantum mechanical effects of the nanoscale can be used to design and control nanoscale structures.

self-assembly

self-assembly: the process by which nanoscale structures form spontaneously. This is a key concept in nanotechnology, where the self-assembly of nanoscale structures can be used to design and control nanoscale structures.



2016 Nobel Prize in Chemistry: for the discovery of graphene

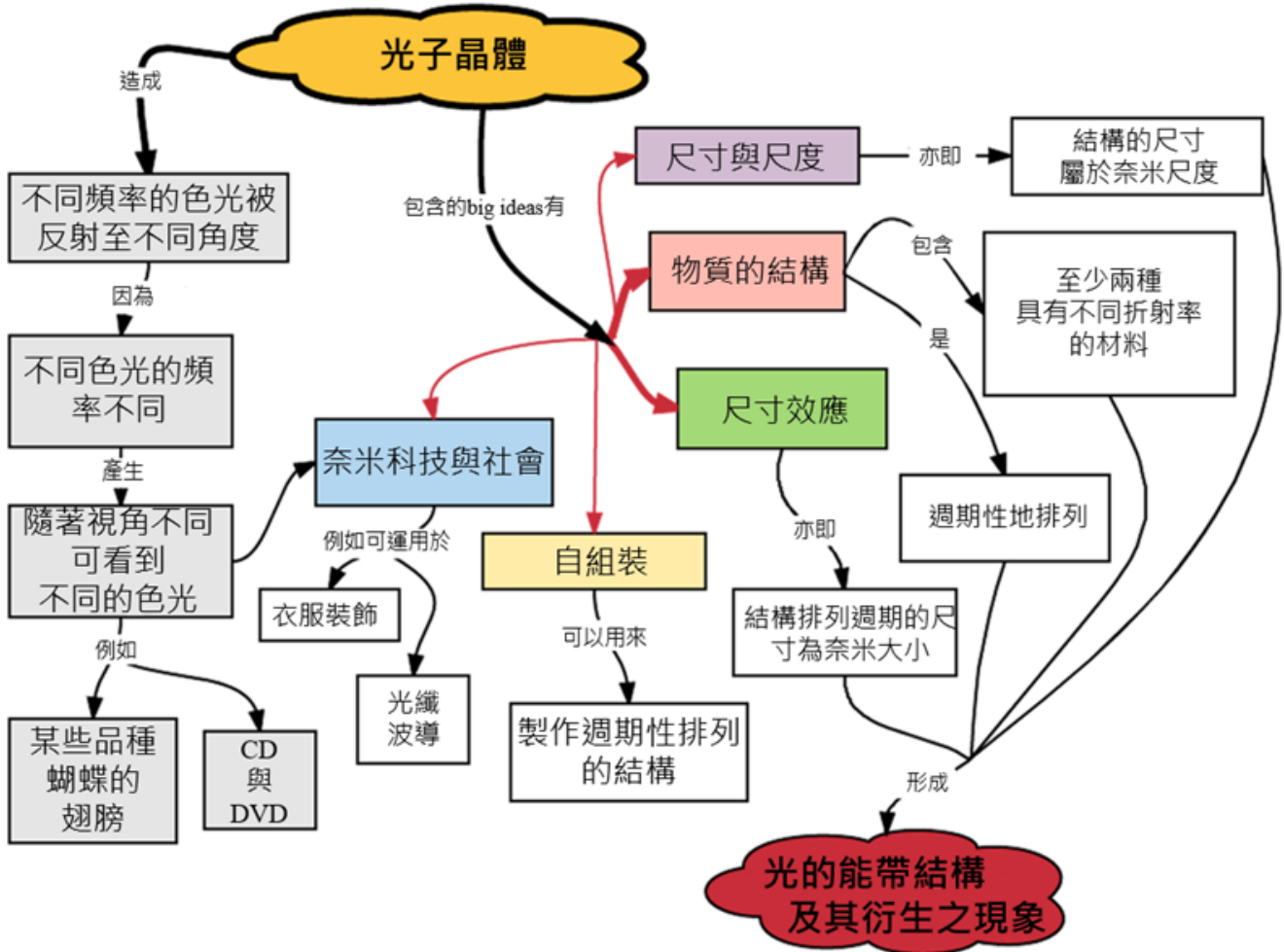
tools & instrumentation

tools & instrumentation: the study of the tools and instrumentation used in nanotechnology. This is a key concept in nanotechnology, where the tools and instrumentation used in nanotechnology can be used to design and control nanoscale structures.

models & simulations

models & simulations: the study of the models and simulations used in nanotechnology. This is a key concept in nanotechnology, where the models and simulations used in nanotechnology can be used to design and control nanoscale structures.

big ideas
 big ideas
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項目	説明	詳細説明
項目1		説明1
項目2	説明2	詳細説明2
項目3	説明3	詳細説明3
項目4	説明4	詳細説明4

この表は、2つの列と3つの行で構成されています。各セルには、項目名、説明、および詳細説明が記載されています。

2つの列と3つの行の表を作成してください。

	説明	詳細説明
項目1		説明1
項目2	説明2 項目... 項目...	詳細説明2 (項目名)
項目3	説明3 (RGB)	詳細説明3



1 Editor abcdef –

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1 Autumn & Peattie, 2002

作用機制	提出假設者	反駁者	反駁之實驗證據
分泌黏液(glue)	N/A	Wagler, 1830; Simmernacher, 1884	壁虎並沒有任何分泌的腺體，不可能分泌黏液
真空吸引(suction, 吸盤)	Simmernacher, 1884	Dellit, 1934	在高真空環境下，壁虎仍具有黏著能力
靜電吸引力 (electrostatics attraction)	Schmidt, 1904	Dellit, 1934	壁虎能在充滿電荷的環境中進行黏著，不受環境電荷的影響
摩擦力(friction)	Hora, 1923; Ruibal & Ernst, 1965	為數眾多	當力垂直於黏著的表面，摩擦力並無法發揮作用，但是壁虎卻可以倒掛在天花板行走。
微交錯作用 (microinterlocking, 俗稱爪力)	Dellit, 1934	Autumn et al., 2000	在表面極光滑的二氧化矽上，壁虎的黏附力仍不受任何影響
毛細作用 (capillary forces)	Hiller, 1968; Huber et al., 2005	Autumn et al., 2002 Arzt, 2006	壁虎的黏附不受吸附表面的化學性質之影響，並且也不受到環境濕度的限制。 壁虎的足底有極高疏水性，極高疏水表面之間，毛細作用中關鍵的毛細橋樑無法成形，但是壁虎卻仍可以黏附在極高疏水表面
凡得瓦力(完全因素)	Stork, 1980; Autumn et al., 2000		

Hiller (1968, 1969, 1975) 認為壁虎的黏附力是由於其足底分泌的黏液所產生。然而，Autumn & Peattie (2002) 指出，壁虎的黏附力並非由黏液產生，而是由其足底特殊的微結構所產生。Hiller 認為壁虎的黏附力是由於其足底分泌的黏液所產生，而 Autumn & Peattie (2002) 則認為壁虎的黏附力是由於其足底特殊的微結構所產生。Hiller 認為壁虎的黏附力是由於其足底分泌的黏液所產生，而 Autumn & Peattie (2002) 則認為壁虎的黏附力是由於其足底特殊的微結構所產生。

van der Waals interaction

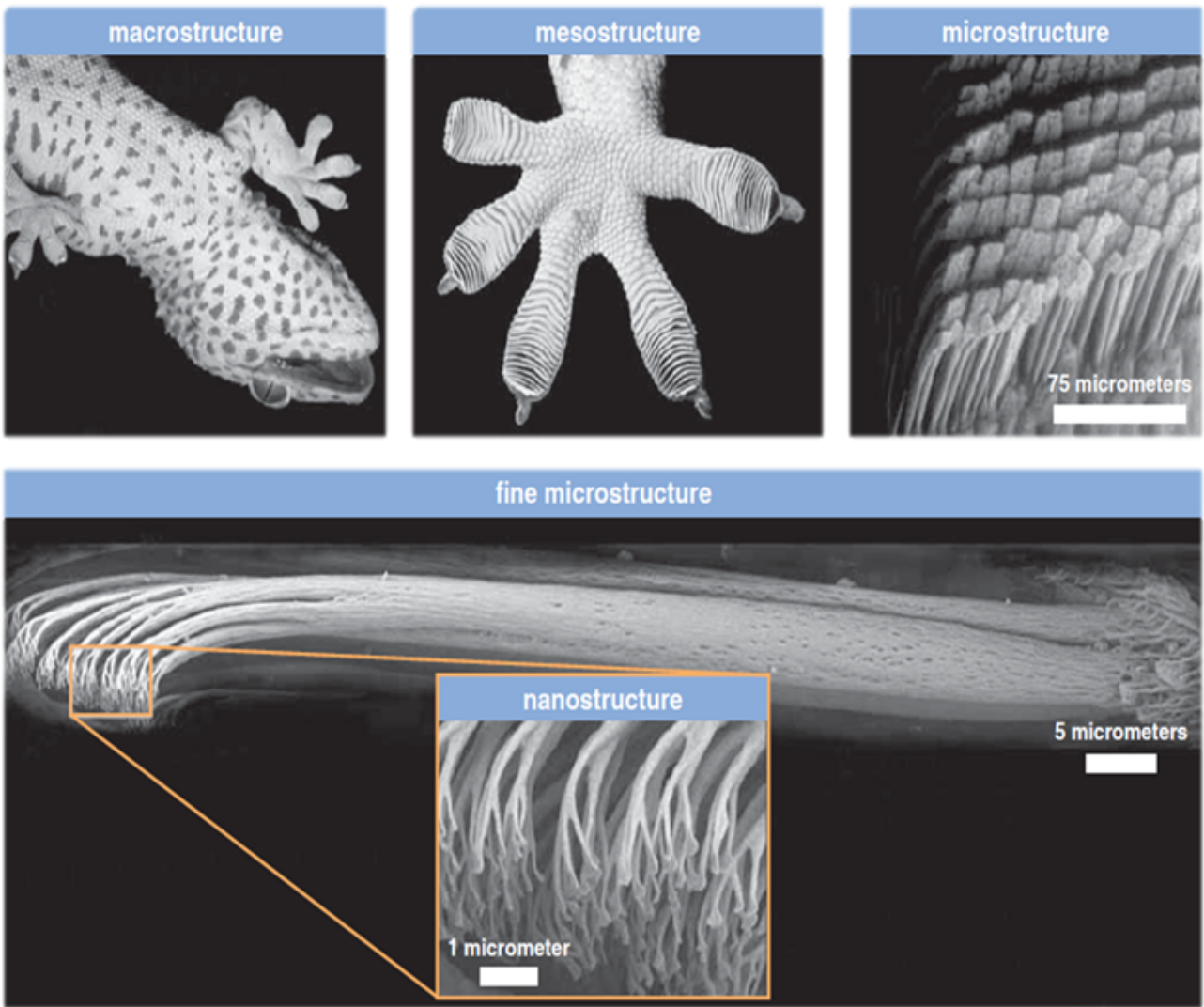
polarizability
 Hiller 1968
 polytetrafluoroethylene, PTFE PTFE
 PTFE Autumn & Peattie, 2002

1965 Ruibal Ernst 1965 scanning electron microscopy, SEM
 setae branch
 100~1000 spatulae 0.1~0.2 μm stalk
 200 nm 0.01 μm 2000
 Kellar Autumn
 Autumn et al., 2000 R
 Hamaker H

$$F = \frac{HR}{6D^2}$$

$H = 10^{-19}$ J $R = 1$ μm $D = 0.2$ nm

400 nN 100~1000 40-400 μN
 Autumn Micro Electro Mechanical Systems MEMS
 200 μN 2002
 Autumn
 Autumn et al., 2002
 -



Autumn, K source: How gecko toes stick. American Scientist 94, 124-132

n

Johnson, Kendall, Roberts 1971 JKR
 Cao et al., 2005 R g

$$F = \frac{3}{2} \gamma \pi R$$

JKR 164~196 nm 200

nm JKR $F \propto R$

請依照下列步驟，將玻璃蓋片裝上蓋玻片，並將蓋玻片蓋上（請依照下列步驟，將玻璃蓋片裝上蓋玻片，並將蓋玻片蓋上）。

1. 將蓋玻片放在載玻片上，並將蓋玻片蓋上，並用橡皮擦輕輕將蓋玻片擦乾。

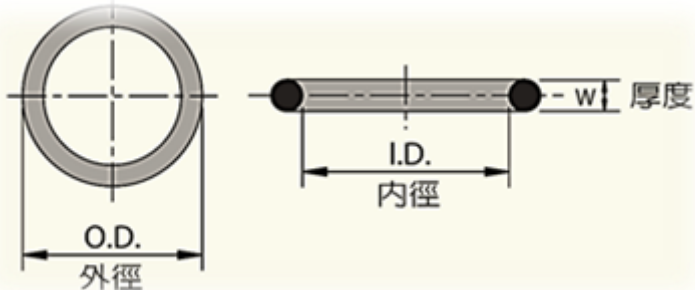
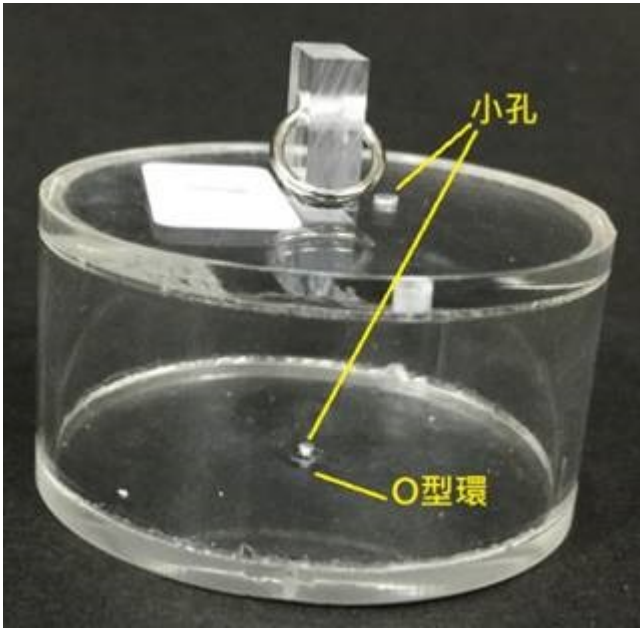
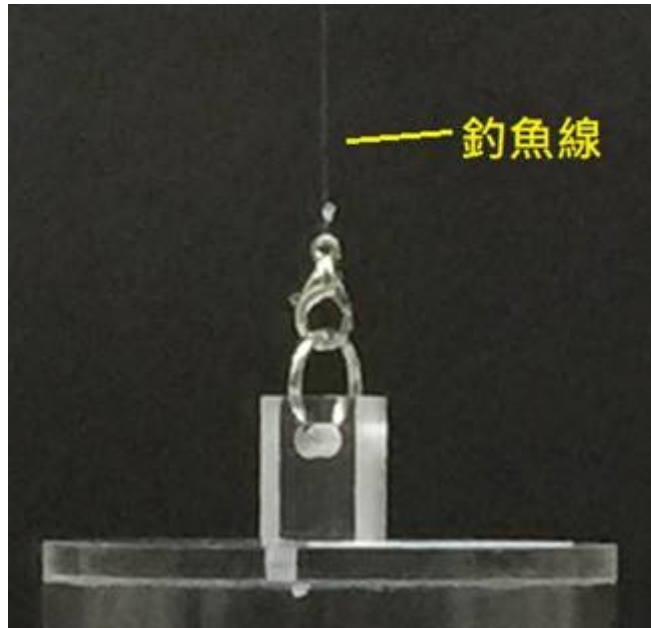


圖3: O型環與小孔的安裝位置





4: □□□□□□□□□□□□□□□□2017□

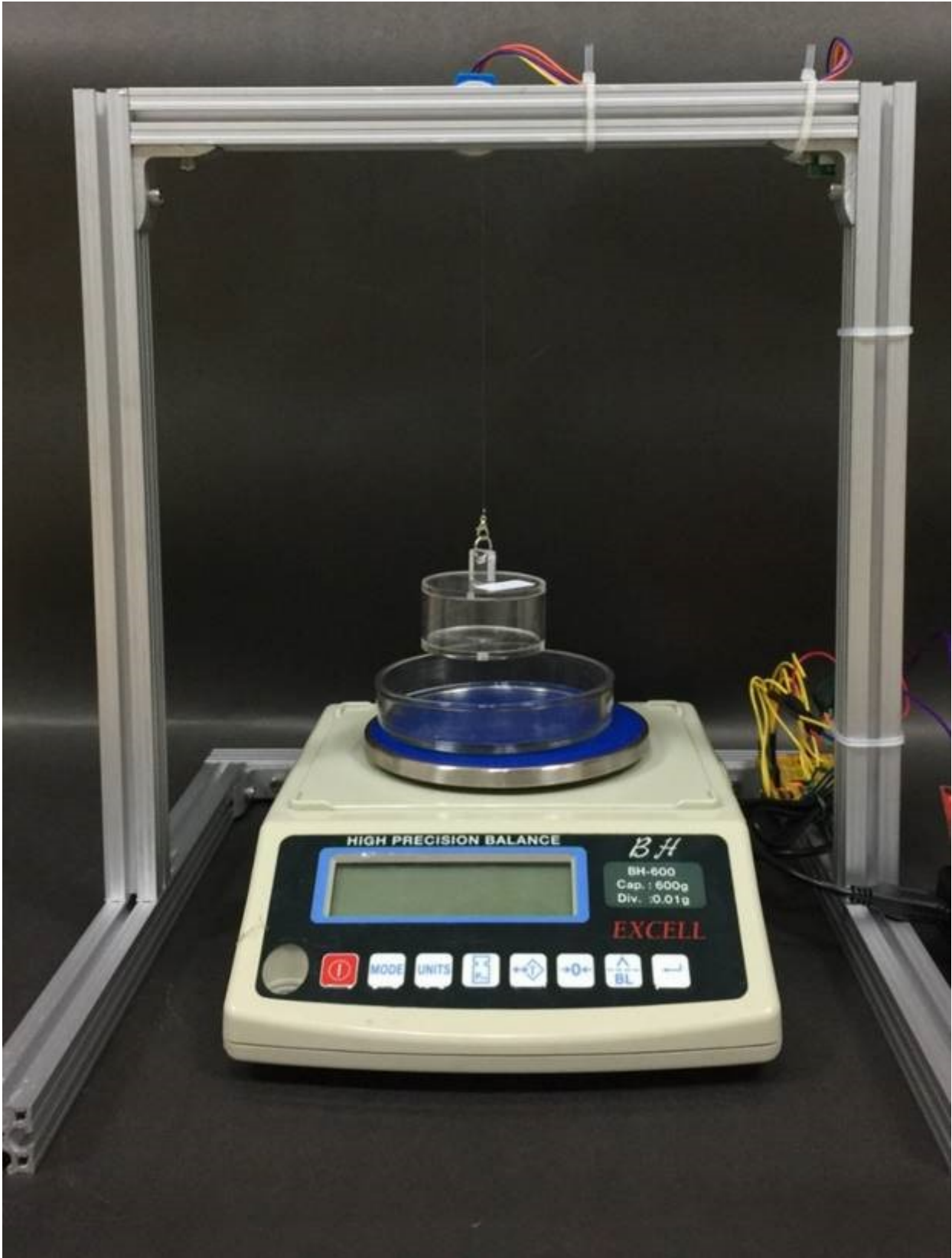


图5: 实验室高精度天平2017

实验室高精度天平2017
实验室高精度天平2017
实验室高精度天平2017

0 1 2 3 mm



圖6: 28BY J-48 步進馬達 (UNL2003) (2017)

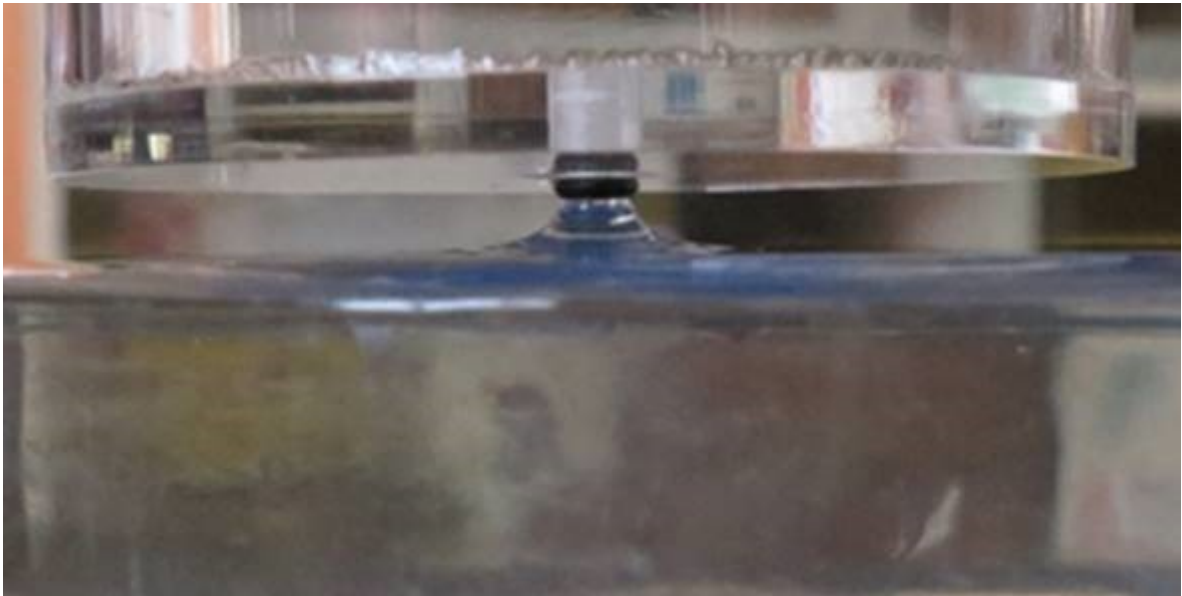


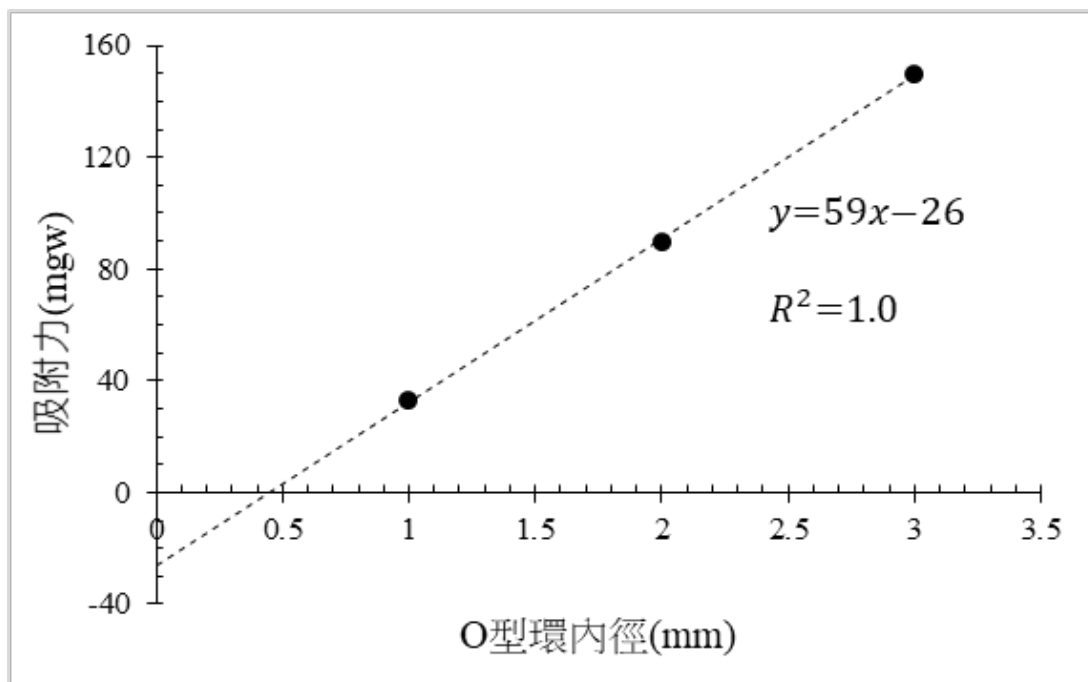
圖6 步進馬達 (2017)

0

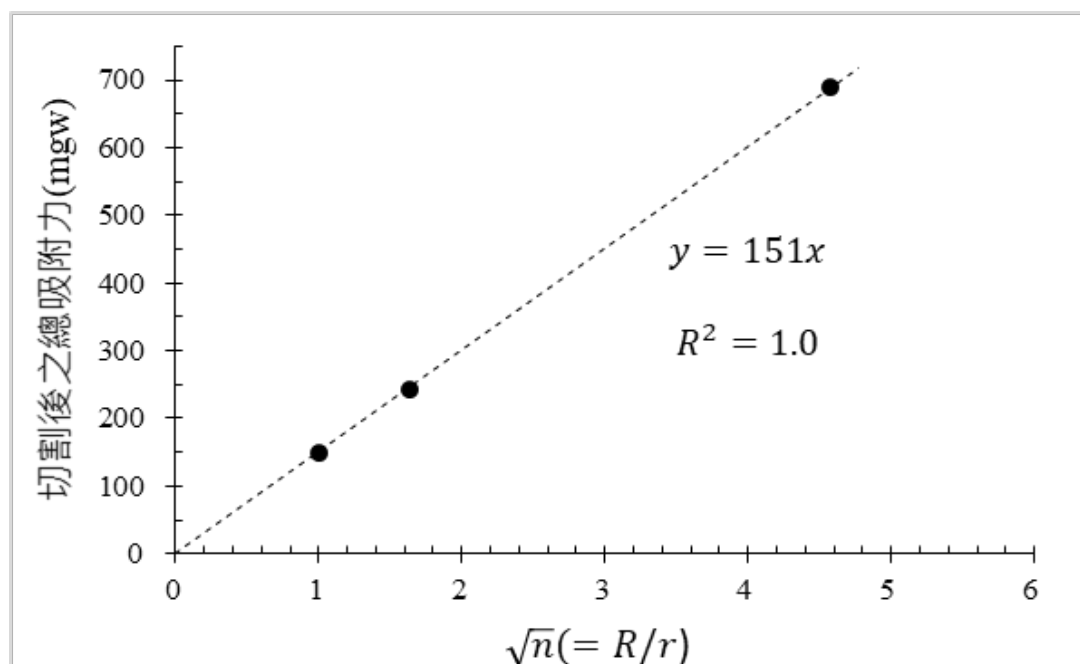
($f \propto r$) 0 0 0

$F' = nf$ $F' \propto \sqrt{n}$ $F' \propto \sqrt{n}$

□□□□



□7□□□□□□□□□□



□8□□□□□□□□ \sqrt{n} □□□□

n □□

□□□□□□□□□□□□□□□□□□□□□□□□□□□□□□□□ $\sqrt{n}(= R/r)$ □□□□□□□□□□□□

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<https://commons.wikimedia.org/w/index.php?curid=87146272>)

▪ [Contact Angle](#)

When a liquid is placed on a solid surface, the contact angle θ is defined as the angle between the tangent to the liquid surface at the point of contact and the solid surface. The contact angle is a measure of the wettability of a solid by a liquid. A contact angle of 0° indicates complete wetting, while a contact angle of 180° indicates complete non-wetting. Contact angles of 10° , 90° , and 120° are commonly used to illustrate different levels of wettability.

(γ_{SL}) f_{AL} f_{SA} f_{SL} $f_{adhesive}$
 adhesive force f_{AL}
 $f_{SL} < f_{SA}$ $\theta < 90^\circ$ f_{AL}
 $f_{SL} > f_{SA}$ $\theta > 90^\circ$

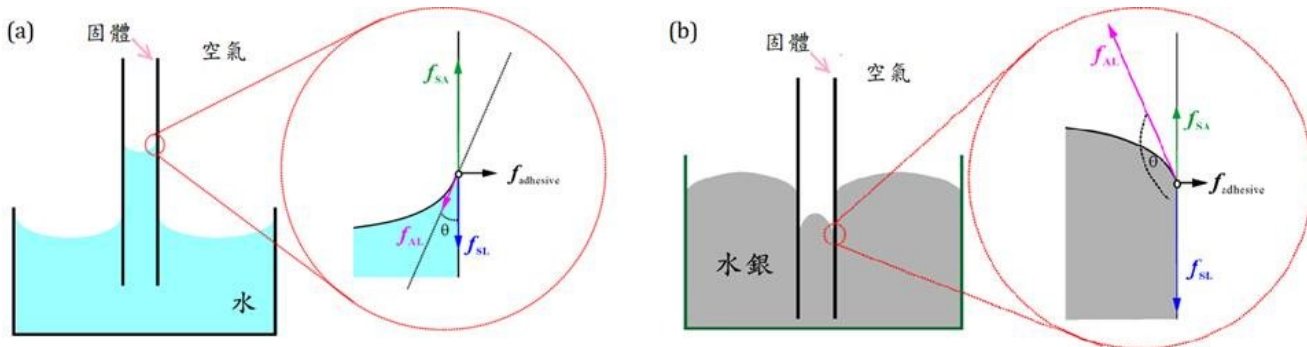


圖4 (a) (b) 2016

(f_{SL}) (f_{SA}) q
 5
 120°

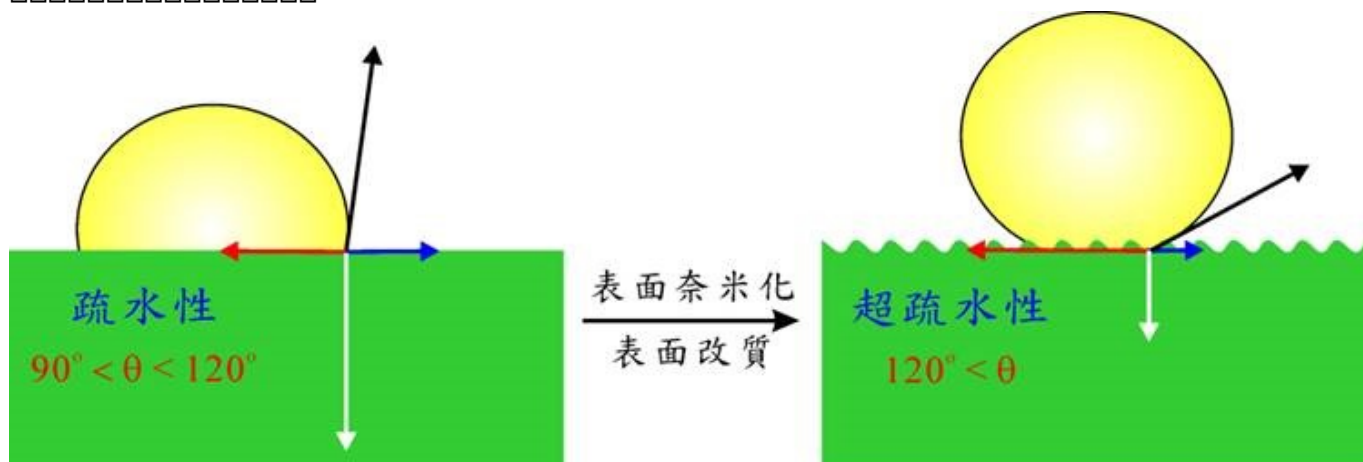
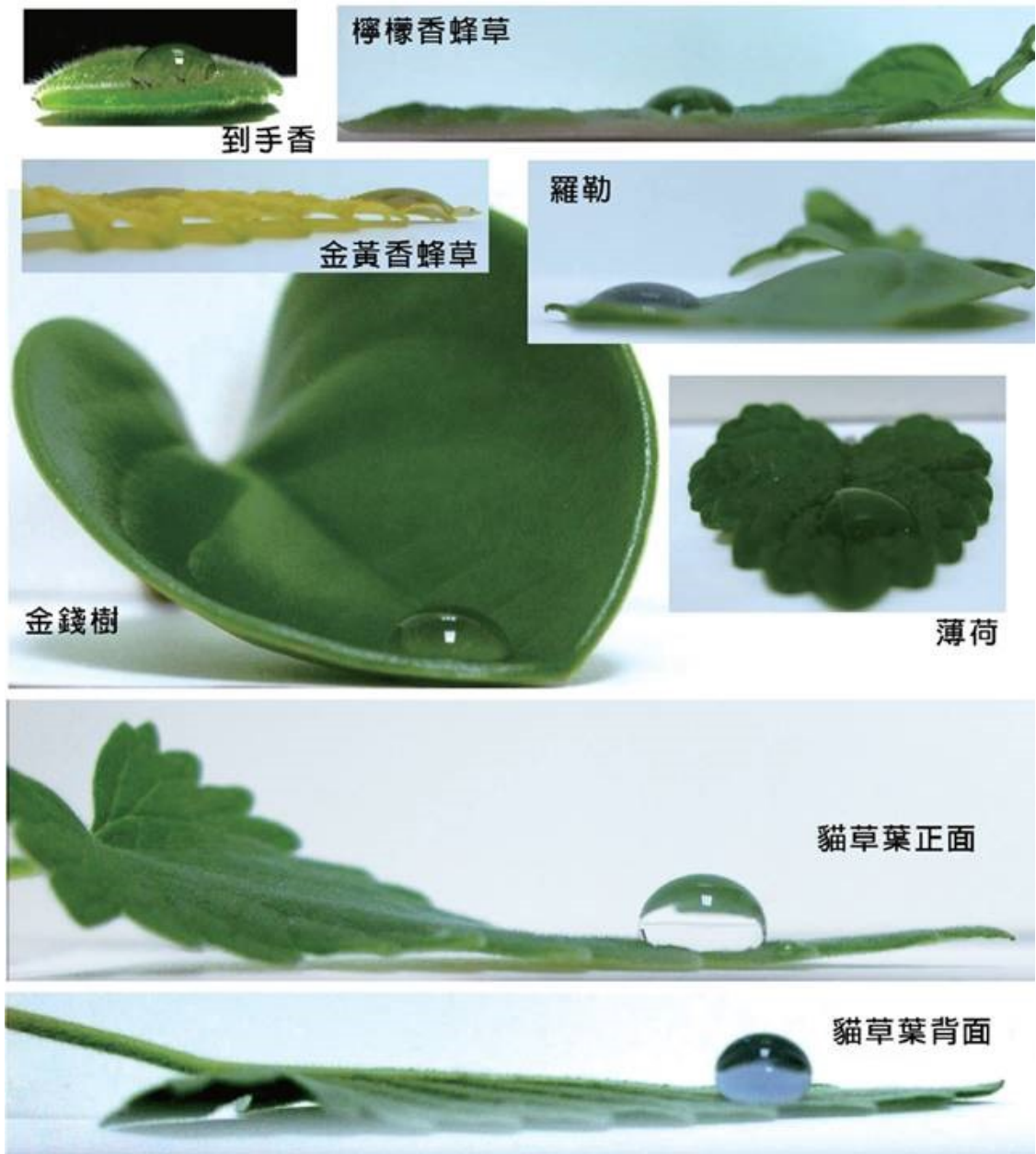


圖5 2016

▪ 超疏水性

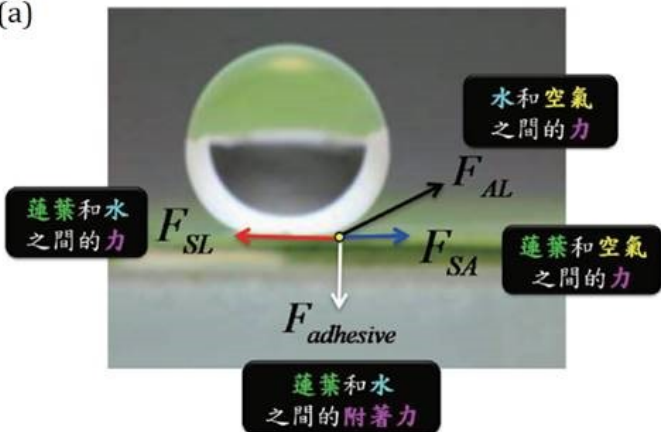
6



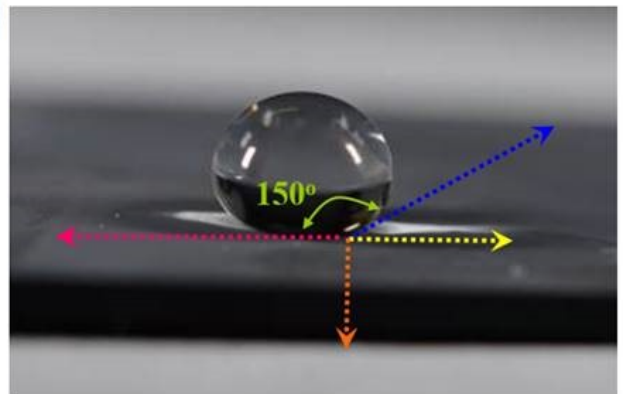
7

4 4 8

(a)

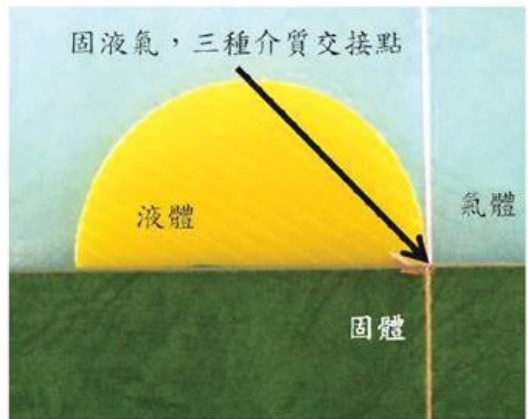


(b)

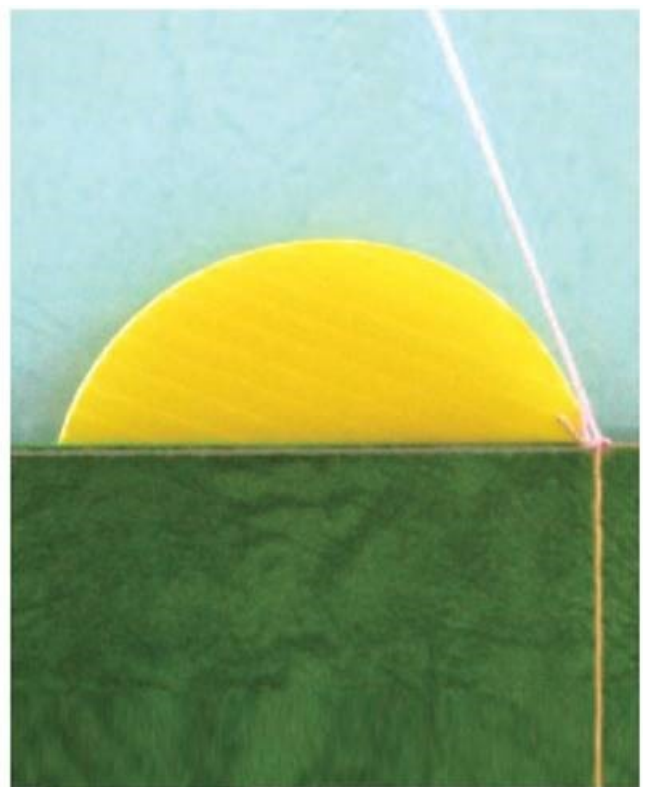
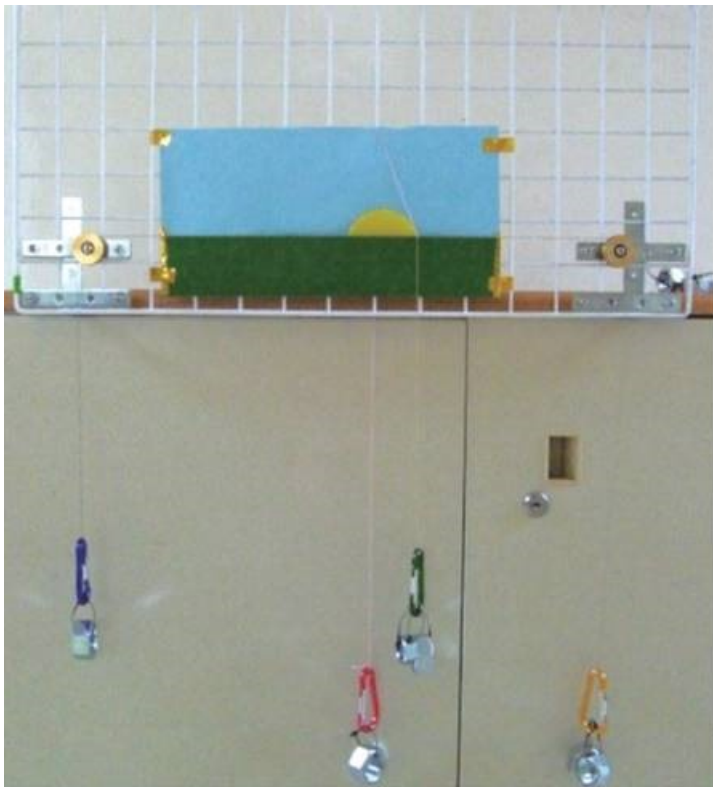


8 (a) (b) 2016

10



11



12

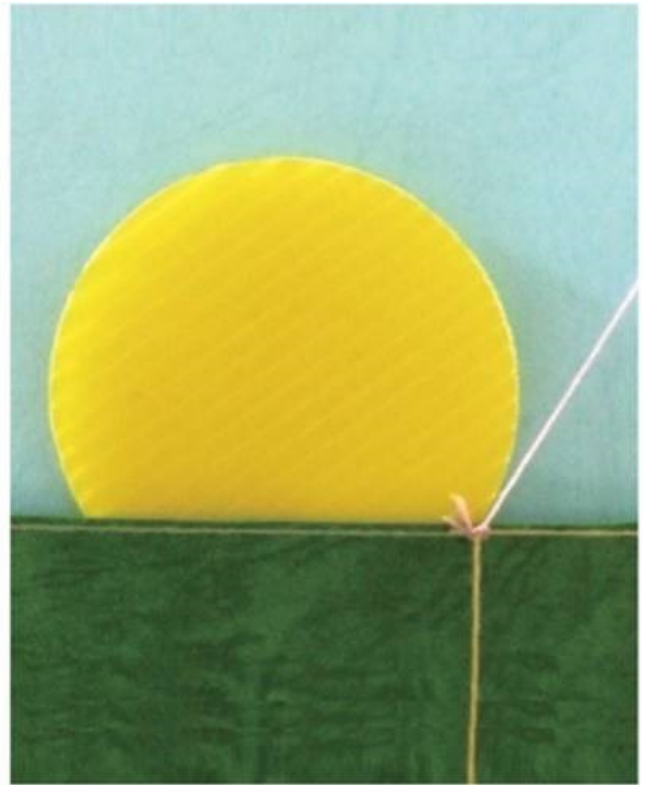


圖13 實驗裝置圖 (2016)

圖1

吊掛的砝碼重量(gw)				接觸角 θ
W_{SL} (固體/液體, f_{SL})	W_{SA} (固體/空氣, f_{SA})	W_{LA} (空氣/液體, f_{LA})	$W_{adhesive}$ (附著力, $f_{adhesive}$)	

實驗步驟

1. 將實驗裝置圖中的黃色半圓形物體，用細線吊起，使其懸掛在金屬架上。
 2. 調整細線的長度，使黃色半圓形物體的直徑與金屬架的寬度相等。
 3. 在金屬架下方，掛上不同重量的砝碼，記錄砝碼的重量。
 4. 觀察黃色半圓形物體與金屬架接觸處的接觸角，並記錄下來。



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